

# InterAtlas Chemical Inc.

Chemwatch: 23004 Version No: 11.1.15.11

Safety Data Sheet according to OSHA HazCom Standard (2012) requirements

Issue Date: 15/04/2021 Print Date: 28/09/2021 S.GHS.USA.EN

### **SECTION 1 Identification**

### **Product Identifier**

Product name	2-ETHYLHEXANOL
Chemical Name	2-ethylhexanol
Synonyms	C8-H18-O; CH3(CH2)3CHC2H5CH2OH; 2-ethyl hexanol; ethylhexanol; 2-ethyl hexylalcohol; 2-ethyl hexyl alcohol; 2-ethyl-1-hexanol; 1-hexanol, 2-ethyl; octyl alcohol; iso-octyl alcohol; 2-EH; isooctanol; iso-octanol; alcohols C7-9-iso, C8 rich (CAS RN: 68526-83-0); alcohols C8-10-iso, C9 rich (CAS RN: 68526-84-1)
Proper shipping name	Environmentally hazardous substance, liquid, n.o.s. (contains 2-ethylhexanol)
Chemical formula	C8-H18-O C8H18O C9H20O
Other means of identification	Not Available
CAS number	104-76-7

### Recommended use of the chemical and restrictions on use

Relevant identified uses

Plasticiser for PVC resins, defoaming agent, wetting agent, organic synthesis, solvent mix for nitrocellulose, paints, lacquers, baking finishes; penetrant for mercerising cotton, textile finishing compounds, plasticisers. Also used in inks, rubber, paper, lubricants, photography and dry cleaning

In perfumery/ flavouring.

# Name, address, and telephone number of the chemical manufacturer, importer, or other responsible party

Registered company name	InterAtlas Chemical Inc.		
Address	63 Church Street, Suite 301, St. Catharines Ontario L2R 3C4 Canada		
Telephone	905-684-9991		
Fax	905-684-4504		
Website	www.interatlaschemical.com		
Email	Not Available		

### **Emergency phone number**

Association / Organisation	CHEMTREC, US
Emergency telephone numbers	800-424-9300
Other emergency telephone numbers	International: +1-703-527-3887

# SECTION 2 Hazard(s) identification

# Classification of the substance or mixture

NFPA 704 diamond



Note: The hazard category numbers found in GHS classification in section 2 of this SDSs are NOT to be used to fill in the NFPA 704 diamond. Blue = Health Red = Fire Yellow = Reactivity White = Special (Oxidizer or water reactive substances)

Classification

Flammable Liquids Category 4, Acute Toxicity (Dermal) Category 4, Skin Corrosion/Irritation Category 2, Serious Eye Damage/Eye Irritation Category 1, Reproductive Toxicity Category 2, Specific Target Organ Toxicity - Single Exposure (Respiratory Tract Irritation) Category 3, Hazardous to the Aquatic Environment Long-Term Hazard Category 2

# Label elements

Issue Date: 15/04/2021 Print Date: 28/09/2021











Signal word	

# Hazard statement(s)

H227	Combustible liquid.
H312	Harmful in contact with skin.
H315	Causes skin irritation.
H318	Causes serious eye damage.
H361	Suspected of damaging fertility or the unborn child.
H335	May cause respiratory irritation.
H411	Toxic to aquatic life with long lasting effects.

# Hazard(s) not otherwise classified

Not Applicable

# Precautionary statement(s) Prevention

P201	Obtain special instructions before use.	
P210	Keep away from heat, hot surfaces, sparks, open flames and other ignition sources. No smoking.	
P235	Keep cool.	
P271	Use only outdoors or in a well-ventilated area.	
P280	Wear protective gloves, protective clothing, eye protection and face protection.	
P261	Avoid breathing mist/vapours/spray.	
P273	Avoid release to the environment.	
P202	Do not handle until all safety precautions have been read and understood.	
P264	Wash all exposed external body areas thoroughly after handling.	

# Precautionary statement(s) Response

P305+P351+P338	IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.			
P308+P313	IF exposed or concerned: Get medical advice/ attention.			
P310	Immediately call a POISON CENTER/doctor/physician/first aider.			
P370+P378	In case of fire: Use alcohol resistant foam or normal protein foam to extinguish.			
P312	Call a POISON CENTER/doctor/physician/first aider/if you feel unwell.			
P391	Collect spillage.			
P302+P352	IF ON SKIN: Wash with plenty of water.			
P304+P340	IF INHALED: Remove person to fresh air and keep comfortable for breathing.			
P332+P313	If skin irritation occurs: Get medical advice/attention.			
P362+P364	Take off contaminated clothing and wash it before reuse.			

# Precautionary statement(s) Storage

P403+P235	Store in a well-ventilated place. Keep cool.	
P405	Store locked up.	
P403+P233	Store in a well-ventilated place. Keep container tightly closed.	

# Precautionary statement(s) Disposal

Dispose of contents/container to authorised hazardous or special waste collection point in accordance with any local regulation.

# **SECTION 3 Composition / information on ingredients**

# Substances

CAS No	%[weight]	Name
104-76-7	>=95	2-ethylhexanol

### Mixtures

See section above for composition of Substances

# **SECTION 4 First-aid measures**

# Description of first aid measures

Chemwatch: **23004**Version No: **11.1.15.11** 

# Page 3 of 13 2-ETHYLHEXANOL

Issue Date: 15/04/2021 Print Date: 28/09/2021

Eye Contact	If this product comes in contact with the eyes:  Immediately hold eyelids apart and flush the eye continuously with running water.  Ensure complete irrigation of the eye by keeping eyelids apart and away from eye and moving the eyelids by occasionally lifting the upper and lower lids.  Continue flushing until advised to stop by the Poisons Information Centre or a doctor, or for at least 15 minutes.  Transport to hospital or doctor without delay.  Removal of contact lenses after an eye injury should only be undertaken by skilled personnel.
Skin Contact	If skin contact occurs:  Immediately remove all contaminated clothing, including footwear.  Flush skin and hair with running water (and soap if available).  Seek medical attention in event of irritation.
Inhalation	<ul> <li>If fumes or combustion products are inhaled remove from contaminated area.</li> <li>Lay patient down. Keep warm and rested.</li> <li>Prostheses such as false teeth, which may block airway, should be removed, where possible, prior to initiating first aid procedures.</li> <li>Apply artificial respiration if not breathing, preferably with a demand valve resuscitator, bag-valve mask device, or pocket mask as trained. Perform CPR if necessary.</li> <li>Transport to hospital, or doctor, without delay.</li> </ul>
Ingestion	<ul> <li>If swallowed do NOT induce vomiting.</li> <li>If vomiting occurs, lean patient forward or place on left side (head-down position, if possible) to maintain open airway and prevent aspiration.</li> <li>Observe the patient carefully.</li> <li>Never give liquid to a person showing signs of being sleepy or with reduced awareness; i.e. becoming unconscious.</li> <li>Give water to rinse out mouth, then provide liquid slowly and as much as casualty can comfortably drink.</li> <li>Seek medical advice.</li> <li>If spontaneous vomiting appears imminent or occurs, hold patient's head down, lower than their hips to help avoid possible aspiration of vomitus.</li> </ul>

## Most important symptoms and effects, both acute and delayed

See Section 11

### Indication of any immediate medical attention and special treatment needed

Any material aspirated during vomiting may produce lung injury. Therefore emesis should not be induced mechanically or pharmacologically. Mechanical means should be used if it is considered necessary to evacuate the stomach contents; these include gastric lavage after endotracheal intubation. If spontaneous vomiting has occurred after ingestion, the patient should be monitored for difficult breathing, as adverse effects of aspiration into the lungs may be delayed up to 48 hours.

To treat poisoning by the higher aliphatic alcohols (up to C7):

- ▶ Gastric lavage with copious amounts of water
- It may be beneficial to instill 60 ml of mineral oil into the stomach.
- Oxygen and artificial respiration as needed.
- Electrolyte balance: it may be useful to start 500 ml. M/6 sodium bicarbonate intravenously but maintain a cautious and conservative attitude toward electrolyte replacement unless shock or severe acidosis threatens.
- ▶ To protect the liver, maintain carbohydrate intake by intravenous infusions of glucose.
- ▶ Haemodialysis if coma is deep and persistent. [GOSSELIN, SMITH HODGE: Clinical Toxicology of Commercial Products, Ed 5)

## BASIC TREATMENT

- ▶ Establish a patent airway with suction where necessary.
- Watch for signs of respiratory insufficiency and assist ventilation as necessary.
- Administer oxygen by non-rebreather mask at 10 to 15 l/min
- Monitor and treat, where necessary, for shock.
- Monitor and treat, where necessary, for pulmonary oedema.
- ► Anticipate and treat, where necessary, for seizures
- DO NOT use emetics. Where ingestion is suspected rinse mouth and give up to 200 ml water (5 ml/kg recommended) for dilution where patient is able to swallow, has a strong gag reflex and does not drool.
- Give activated charcoal.

### ADVANCED TREATMENT

- Consider orotracheal or nasotracheal intubation for airway control in unconscious patient or where respiratory arrest has occurred.
- ▶ Positive-pressure ventilation using a bag-valve mask might be of use.
- Monitor and treat, where necessary, for arrhythmias.
- Start an IV D5W TKO. If signs of hypovolaemia are present use lactated Ringers solution. Fluid overload might create complications.
- If the patient is hypoglycaemic (decreased or loss of consciousness, tachycardia, pallor, dilated pupils, diaphoresis and/or dextrose strip or glucometer readings below 50 mg), give 50% dextrose.
- Hypotension with signs of hypovolaemia requires the cautious administration of fluids. Fluid overload might create complications.
- Drug therapy should be considered for pulmonary oedema.
- Treat seizures with diazepam.
- Proparacaine hydrochloride should be used to assist eye irrigation.

# EMERGENCY DEPARTMENT

- Laboratory analysis of complete blood count, serum electrolytes, BUN, creatinine, glucose, urinalysis, baseline for serum aminotransferases (ALT and AST), calcium, phosphorus and magnesium, may assist in establishing a treatment regime. Other useful analyses include anion and osmolar gaps, arterial blood gases (ABGs), chest radiographs and electrocardiograph.
- Positive end-expiratory pressure (PEEP)-assisted ventilation may be required for acute parenchymal injury or adult respiratory distress syndrome.
- Acidosis may respond to hyperventilation and bicarbonate therapy.
- Haemodialysis might be considered in patients with severe intoxication.
- Consult a toxicologist as necessary. BRONSTEIN, A.C. and CURRANCE, P.L. EMERGENCY CARE FOR HAZARDOUS MATERIALS EXPOSURE: 2nd Ed. 1994

For C8 alcohols and above.

Symptomatic and supportive therapy is advised in managing patients.

# **SECTION 5 Fire-fighting measures**

Issue Date: **15/04/2021**Print Date: **28/09/2021** 

- Alcohol stable foam.
- Dry chemical powder.
- BCF (where regulations permit).
- Carbon dioxide.
- Water spray or fog Large fires only.

### Special hazards arising from the substrate or mixture

Fire Incompatibility

Fire Fighting

▶ Avoid contamination with oxidising agents i.e. nitrates, oxidising acids, chlorine bleaches, pool chlorine etc. as ignition may result

# Special protective equipment and precautions for fire-fighters

- Alert Fire Brigade and tell them location and nature of hazard.
   Wear full body protective clothing with breathing apparatus.
  - Prevent, by any means available, spillage from entering drains or water course.
  - Use water delivered as a fine spray to control fire and cool adjacent area
  - Avoid spraying water onto liquid pools.
  - ▶ DO NOT approach containers suspected to be hot.
  - Cool fire exposed containers with water spray from a protected location.
  - If safe to do so, remove containers from path of fire.

#### Combustible

- ▶ Slight fire hazard when exposed to heat or flame.
- ▶ Heating may cause expansion or decomposition leading to violent rupture of containers.
- On combustion, may emit toxic fumes of carbon monoxide (CO).
- Fire/Explosion Hazard May emit acrid smoke.
  - ▶ Mists containing combustible materials may be explosive.

Combustion products include:

Environmental hazard contain spillage

carbon dioxide (CO2)

other pyrolysis products typical of burning organic material.

### **SECTION 6 Accidental release measures**

# Personal precautions, protective equipment and emergency procedures

See section 8

### **Environmental precautions**

See section 12

### Methods and material for containment and cleaning up

Minor Spills	Clean up all spills immediately.     Avoid breathing vapours and contact with skin and eyes.     Control personal contact with the substance, by using protective equipment.     Contain and absorb spill with sand, earth, inert material or vermiculite.     Wipe up.     Place in a suitable, labelled container for waste disposal.
Major Spills	Environmental hazard - contain spillage.  Moderate hazard.  Clear area of personnel and move upwind.  Alert Fire Brigade and tell them location and nature of hazard.  Wear breathing apparatus plus protective gloves.  Prevent, by any means available, spillage from entering drains or water course.  No smoking, naked lights or ignition sources.  Increase ventilation.  Stop leak if safe to do so.  Contain spill with sand, earth or vermiculite.  Collect recoverable product into labelled containers for recycling.  Absorb remaining product with sand, earth or vermiculite.  Collect solid residues and seal in labelled drums for disposal.

Personal Protective Equipment advice is contained in Section 8 of the SDS.

# **SECTION 7 Handling and storage**

Safe handling

# Precautions for safe handling

DO NOT allow clothing wet with material to stay in contact with skin

If contamination of drains or waterways occurs, advise emergency services

Avoid all personal contact, including inhalation.

Wash area and prevent runoff into drains.

- Wear protective clothing when risk of exposure occurs
- ▶ Use in a well-ventilated area.
- Prevent concentration in hollows and sumps.
- DO NOT enter confined spaces until atmosphere has been checked.
- Avoid smoking, naked lights or ignition sources.
  - Avoid contact with incompatible materials.
  - ► When handling, **DO NOT** eat, drink or smoke.
  - Keep containers securely sealed when not in use.
  - Avoid physical damage to containers.
  - Always wash hands with soap and water after handling.
  - Work clothes should be laundered separately.
  - Use good occupational work practice.
  - ▶ Observe manufacturer's storage and handling recommendations contained within this SDS.

Issue Date: 15/04/2021 Print Date: 28/09/2021

- Atmosphere should be regularly checked against established exposure standards to ensure safe working conditions. Store in original containers. ► Keep containers securely sealed.
  - ▶ Store in a cool, dry, well-ventilated area. Other information

    - Store away from incompatible materials and foodstuff containers.
    - Protect containers against physical damage and check regularly for leaks.
    - ▶ Observe manufacturer's storage and handling recommendations contained within this SDS.

### Conditions for safe storage, including any incompatibilities

### Suitable container

- ▶ Glass container is suitable for laboratory quantities
- Metal can or drum
- Packaging as recommended by manufacturer.
- ► Check all containers are clearly labelled and free from leaks.

### Alcohols

- are incompatible with strong acids, acid chlorides, acid anhydrides, oxidising and reducing agents.
- reacts, possibly violently, with alkaline metals and alkaline earth metals to produce hydrogen

### Storage incompatibility

- react with strong acids, strong caustics, aliphatic amines, isocyanates, acetaldehyde, benzoyl peroxide, chromic acid, chromium oxide, dialkylzincs, dichlorine oxide, ethylene oxide, hypochlorous acid, isopropyl chlorocarbonate, lithium tetrahydroaluminate, nitrogen dioxide, pentafluoroguanidine, phosphorus halides, phosphorus pentasulfide, tangerine oil, triethylaluminium, triisobutylaluminium
- should not be heated above 49 deg. C. when in contact with aluminium equipment

Secondary alcohols and some branched primary alcohols may produce potentially explosive peroxides after exposure to light and/ or heat.

### SECTION 8 Exposure controls / personal protection

### **Control parameters**

Occupational Exposure Limits (OEL)

### **INGREDIENT DATA**

Not Available

#### Emergency Limits

Ingredient	TEEL-1 TEEL-2			TEEL-3
2-ethylhexanol	Not Available	Not Available		Not Available
Ingredient	Original IDLH		Revised IDLH	
2-ethylhexanol	Not Available		Not Available	

# Occupational Exposure Banding

Ingredient	Occupational Exposure Band Rating	Occupational Exposure Band Limit	
2-ethylhexanol	E	≤ 0.1 ppm	
Notes:	Occupational exposure banding is a process of assigning chemicals into specific categories or bands based on a chemical's potency and the adverse health outcomes associated with exposure. The output of this process is an occupational exposure band (OEB), which corresponds to a range of exposure concentrations that are expected to protect worker health.		

# **Exposure controls**

Engineering controls are used to remove a hazard or place a barrier between the worker and the hazard. Well-designed engineering controls can be highly effective in protecting workers and will typically be independent of worker interactions to provide this high level of protection. The basic types of engineering controls are:

Process controls which involve changing the way a job activity or process is done to reduce the risk.

Enclosure and/or isolation of emission source which keeps a selected hazard "physically" away from the worker and ventilation that strategically "adds" and "removes" air in the work environment. Ventilation can remove or dilute an air contaminant if designed properly. The design of a ventilation system must match the particular process and chemical or contaminant in use.

Employers may need to use multiple types of controls to prevent employee overexposure.

Local exhaust ventilation usually required. If risk of overexposure exists, wear approved respirator. Correct fit is essential to obtain adequate protection. Supplied-air type respirator may be required in special circumstances. Correct fit is essential to ensure adequate protection. An approved self contained breathing apparatus (SCBA) may be required in some situations.

Provide adequate ventilation in warehouse or closed storage area. Air contaminants generated in the workplace possess varying "escape" velocities which, in turn, determine the "capture velocities" of fresh circulating air required to effectively remove the contaminant.

# Appropriate engineering

Type of Contaminant:	Air Speed:
solvent, vapours, degreasing etc., evaporating from tank (in still air).	0.25-0.5 m/s (50-100 f/min.)
aerosols, fumes from pouring operations, intermittent container filling, low speed conveyer transfers, welding, spray drift, plating acid fumes, pickling (released at low velocity into zone of active generation)	0.5-1 m/s (100-200 f/min.)
direct spray, spray painting in shallow booths, drum filling, conveyer loading, crusher dusts, gas discharge (active generation into zone of rapid air motion)	1-2.5 m/s (200-500 f/min.)
grinding, abrasive blasting, tumbling, high speed wheel generated dusts (released at high initial velocity into zone of very high rapid air motion).	2.5-10 m/s (500-2000 f/min.)

Within each range the appropriate value depends on:

Lower end of the range	Upper end of the range
1: Room air currents minimal or favourable to capture	1: Disturbing room air currents
2: Contaminants of low toxicity or of nuisance value only.	2: Contaminants of high toxicity
3: Intermittent, low production.	3: High production, heavy use
4: Large hood or large air mass in motion	4: Small hood-local control only

Issue Date: **15/04/2021**Print Date: **28/09/2021** 

Simple theory shows that air velocity falls rapidly with distance away from the opening of a simple extraction pipe. Velocity generally decreases with the square of distance from the extraction point (in simple cases). Therefore the air speed at the extraction point should be adjusted, accordingly, after reference to distance from the contaminating source. The air velocity at the extraction fan, for example, should be a minimum of 1-2 m/s (200-400 f/min) for extraction of solvents generated in a tank 2 meters distant from the extraction point. Other mechanical considerations, producing performance deficits within the extraction apparatus, make it essential that theoretical air velocities are multiplied by factors of 10 or more when extraction systems are installed or used.

#### Personal protection













# Eye and face protection

Safety glasses with side shields.

- Chemical goggles
- Contact lenses may pose a special hazard; soft contact lenses may absorb and concentrate irritants. A written policy document, describing the wearing of lenses or restrictions on use, should be created for each workplace or task. This should include a review of lens absorption and adsorption for the class of chemicals in use and an account of injury experience. Medical and first-aid personnel should be trained in their removal and suitable equipment should be readily available. In the event of chemical exposure, begin eye irrigation immediately and remove contact lens as soon as practicable. Lens should be removed at the first signs of eye redness or irritation lens should be removed in a clean environment only after workers have washed hands thoroughly. [CDC NIOSH Current Intelligence Bulletin 59], [AS/NZS 1336 or national equivalent]

#### Skin protection

Hands/feet protection

See Hand protection below

- ► Wear chemical protective gloves, e.g. PVC.
- ▶ Wear safety footwear or safety gumboots, e.g. Rubber

#### NOTE:

- The material may produce skin sensitisation in predisposed individuals. Care must be taken, when removing gloves and other protective equipment, to avoid all possible skin contact.
- Contaminated leather items, such as shoes, belts and watch-bands should be removed and destroyed.

The selection of suitable gloves does not only depend on the material, but also on further marks of quality which vary from manufacturer to manufacturer. Where the chemical is a preparation of several substances, the resistance of the glove material can not be calculated in advance and has therefore to be checked prior to the application.

The exact break through time for substances has to be obtained from the manufacturer of the protective gloves and has to be observed when making a final choice.

Personal hygiene is a key element of effective hand care. Gloves must only be worn on clean hands. After using gloves, hands should be washed and dried thoroughly. Application of a non-perfumed moisturiser is recommended.

Suitability and durability of glove type is dependent on usage. Important factors in the selection of gloves include:

- frequency and duration of contact,
- · chemical resistance of glove material,
- glove thickness and
- dexterity

Select gloves tested to a relevant standard (e.g. Europe EN 374, US F739, AS/NZS 2161.1 or national equivalent).

· When prolonged or frequently repeated contact may occur, a glove with a protection class of 5 or higher (breakthrough time greater than 240 minutes according to EN 374, AS/NZS 2161.10.1 or national equivalent) is recommended.

When only brief contact is expected, a glove with a protection class of 3 or higher (breakthrough time greater than 60 minutes according to EN 374, AS/NZS 2161.10.1 or national equivalent) is recommended.

- Some glove polymer types are less affected by movement and this should be taken into account when considering gloves for long-term use.
- · Contaminated gloves should be replaced.

As defined in ASTM F-739-96 in any application, gloves are rated as:

- · Excellent when breakthrough time > 480 min
- Good when breakthrough time > 20 min
- Fair when breakthrough time < 20 min
- Poor when glove material degrades

For general applications, gloves with a thickness typically greater than 0.35 mm, are recommended.

It should be emphasised that glove thickness is not necessarily a good predictor of glove resistance to a specific chemical, as the permeation efficiency of the glove will be dependent on the exact composition of the glove material. Therefore, glove selection should also be based on consideration of the task requirements and knowledge of breakthrough times.

Glove thickness may also vary depending on the glove manufacturer, the glove type and the glove model. Therefore, the manufacturers' technical data should always be taken into account to ensure selection of the most appropriate glove for the task.

Note: Depending on the activity being conducted, gloves of varying thickness may be required for specific tasks. For example:

- Thinner gloves (down to 0.1 mm or less) may be required where a high degree of manual dexterity is needed. However, these gloves are only likely to give short duration protection and would normally be just for single use applications, then disposed of.
- Thicker gloves (up to 3 mm or more) may be required where there is a mechanical (as well as a chemical) risk i.e. where there is abrasion or puncture potential

Gloves must only be worn on clean hands. After using gloves, hands should be washed and dried thoroughly. Application of a non-perfumed moisturiser is recommended.

### **Body protection**

See Other protection below

# Other protection

- Overalls.
- P.V.C apron.Barrier cream.
- ► Skin cleansing cream.
- Eye wash unit.

# Recommended material(s)

### **GLOVE SELECTION INDEX**

Glove selection is based on a modified presentation of the:

"Forsberg Clothing Performance Index".

The effect(s) of the following substance(s) are taken into account in the *computer- qenerated* selection:

2-ETHYLHEXANOL

Material	СРІ
BUTYL	Α

### Respiratory protection

Type A Filter of sufficient capacity. (AS/NZS 1716 & 1715, EN 143:2000 & 149:2001, ANSI Z88 or national equivalent)

Selection of the Class and Type of respirator will depend upon the level of breathing zone contaminant and the chemical nature of the contaminant. Protection Factors (defined as the ratio of contaminant outside and inside the mask) may also be important.

				4
Required minimum protection factor	Maximum gas/vapour concentration present in air p.p.m. (by volume)	Half-face Respirator	Full-Face Respirator	

Issue Date: **15/04/2021**Print Date: **28/09/2021** 

NEOPRENE	A
PVA	A
VITON	Α

- \* CPI Chemwatch Performance Index
- A: Best Selection
- B: Satisfactory; may degrade after 4 hours continuous immersion
- C: Poor to Dangerous Choice for other than short term immersion

**NOTE**: As a series of factors will influence the actual performance of the glove, a final selection must be based on detailed observation. -

\* Where the glove is to be used on a short term, casual or infrequent basis, factors such as "feel" or convenience (e.g. disposability), may dictate a choice of gloves which might otherwise be unsuitable following long-term or frequent use. A qualified practitioner should be consulted.

up to 10	1000	A-AUS / Class1	-
up to 50	1000	-	A-AUS / Class 1
up to 50	5000	Airline *	-
up to 100	5000	-	A-2
up to 100	10000	-	A-3
100+			Airline**

- \* Continuous Flow \*\* Continuous-flow or positive pressure demand A(All classes) = Organic vapours, B AUS or B1 = Acid gasses, B2 = Acid gas or hydrogen cyanide(HCN), B3 = Acid gas or hydrogen cyanide(HCN), E = Sulfur dioxide(SO2), G = Agricultural chemicals, K = Ammonia(NH3), Hg = Mercury, NO = Oxides of nitrogen, MB = Methyl bromide, AX = Low boiling point organic compounds(below 65 degC)
  - Cartridge respirators should never be used for emergency ingress or in areas of unknown vapour concentrations or oxygen content.
  - The wearer must be warned to leave the contaminated area immediately on detecting any odours through the respirator. The odour may indicate that the mask is not functioning properly, that the vapour concentration is too high, or that the mask is not properly fitted. Because of these limitations, only restricted use of cartridge respirators is considered appropriate.
  - Cartridge performance is affected by humidity. Cartridges should be changed after 2 hr of continuous use unless it is determined that the humidity is less than 75%, in which case, cartridges can be used for 4 hr. Used cartridges should be discarded daily, regardless of the length of time used

# **SECTION 9 Physical and chemical properties**

# Information on basic physical and chemical properties

Appearance	Clear colourless liquid with mild, sweet, slightly floral odour. Slightly soluble in water (1379 mg/l). Miscible with most organic solvents.			
Physical state	Liquid	Relative density (Water = 1)	0.83	
Odour	Not Available	Partition coefficient n-octanol / water	Not Available	
Odour threshold	Not Available	Auto-ignition temperature (°C)	288	
pH (as supplied)	Not Applicable	Decomposition temperature	Not Available	
Melting point / freezing point (°C)	- 76	Viscosity (cSt)	Not Available	
Initial boiling point and boiling range (°C)	183.5	Molecular weight (g/mol)	130.26	
Flash point (°C)	81.1	Taste	Not Available	
Evaporation rate	0.01 BuAc=1	Explosive properties	Not Available	
Flammability	Combustible.	Oxidising properties	Not Available	
Upper Explosive Limit (%)	9.7	Surface Tension (dyn/cm or mN/m)	Not Available	
Lower Explosive Limit (%)	0.88	Volatile Component (%vol)	100	
Vapour pressure (kPa)	0.36 @ 20 C.	Gas group	Not Available	
Solubility in water	Partly miscible	pH as a solution (%)	7 (approx.)	
Vapour density (Air = 1)	4.5	VOC g/L	Not Available	

# **SECTION 10 Stability and reactivity**

Reactivity	See section 7
Chemical stability	<ul> <li>Unstable in the presence of incompatible materials.</li> <li>Product is considered stable.</li> <li>Hazardous polymerisation will not occur.</li> </ul>
Possibility of hazardous reactions	See section 7
Conditions to avoid	See section 7
Incompatible materials	See section 7
Hazardous decomposition products	See section 5

# **SECTION 11 Toxicological information**

# Information on toxicological effects

Inhaled

The material can cause respiratory irritation in some persons. The body's response to such irritation can cause further lung damage. Inhalation of vapours may cause drowsiness and dizziness. This may be accompanied by sleepiness, reduced alertness, loss of reflexes, lack of co-ordination, and vertigo.

Inhalation of vapours or aerosols (mists, fumes), generated by the material during the course of normal handling, may be damaging to the health

Chemwatch: **23004** Version No: **11.1.15.11**  Page 8 of 13

2-ETHYLHEXANOL

Issue Date: **15/04/2021**Print Date: **28/09/2021** 

	l '	s, drowsiness, muscle weakness and delirium, central depression, coma, and failure, as well as low blood pressure and irregular heart rhythms, may
Ingestion	Swallowing of the liquid may cause aspiration into the lungs with the ri- (ICSC13733)  Accidental ingestion of the material may be damaging to the health of of the overexposure to non-ring alcohols causes nervous system symptoms. giddiness, confusion, delirium and coma.	the individual.
Skin Contact	cause contact dermatitis which is characterised by redness, swelling a Repeated exposure may cause skin cracking, flaking or drying followin Most liquid alcohols appear to act as primary skin irritants in humans. I man.  Open cuts, abraded or irritated skin should not be exposed to this mate	wing direct contact or after a delay of some time. Repeated exposure can nd blistering.  g normal handling and use.  Significant percutaneous absorption occurs in rabbits but not apparently in erial esions, may produce systemic injury with harmful effects. Examine the skin
Eye	If applied to the eyes, this material causes severe eye damage.	
Chronic	Chronic  Chronic  Chronic  Chronic  Chronic  Chronic  Chronic  Long-term exposure to respiratory irritants may result in airways disease, involving difficulty breathing and related whole-body problems.  Based on experience with animal studies, exposure to the material may result in toxic effects to the development of the foetus, at levels which not cause significant toxic effects to the mother.  Substance accumulation, in the human body, may occur and may cause some concern following repeated or long-term occupational exposurations. There is limited evidence that, skin contact with this product is more likely to cause a sensitisation reaction in some persons compared to the general population.	
	TOWNER	IDDITATION

2-ethylhexanol

TOXICITY	IRRITATION
Dermal (rabbit) LD50: 1970 mg/kg <sup>[2]</sup>	Eye (rabbit): 20 mg/24h - moderate
Oral(Guinea) LD50; 600 mg/kg <sup>[2]</sup>	Eye (rabbit): 4.17 mg - SEVERE
	Eye: adverse effect observed (irritating) <sup>[1]</sup>
	Skin (rabbit): 415 mg (open)-mild
	Skin (rabbit): 500 mg/24h-moderate
	Skin: adverse effect observed (irritating) <sup>[1]</sup>

Legend:

1. Value obtained from Europe ECHA Registered Substances - Acute toxicity 2.\* Value obtained from manufacturer's SDS. Unless otherwise specified data extracted from RTECS - Register of Toxic Effect of chemical Substances

2-ETHYLHEXANOL

Asthma-like symptoms may continue for months or even years after exposure to the material ends. This may be due to a non-allergic condition known as reactive airways dysfunction syndrome (RADS) which can occur after exposure to high levels of highly irritating compound. Main criteria for diagnosing RADS include the absence of previous airways disease in a non-atopic individual, with sudden onset of persistent asthma-like symptoms within minutes to hours of a documented exposure to the irritant. Other criteria for diagnosis of RADS include a reversible airflow pattern on lung function tests, moderate to severe bronchial hyperreactivity on methacholine challenge testing, and the lack of minimal lymphocytic inflammation, without eosinophilia. RADS (or asthma) following an irritating inhalation is an infrequent disorder with rates related to the concentration of and duration of exposure to the irritating substance. On the other hand, industrial bronchitis is a disorder that occurs as a result of exposure due to high concentrations of irritating substance (often particles) and is completely reversible after exposure ceases. The disorder is characterized by difficulty breathing, cough and mucus production.

The material may produce severe irritation to the eye causing pronounced inflammation. Repeated or prolonged exposure to irritants may produce conjunctivitis.

The material may cause skin irritation after prolonged or repeated exposure and may produce on contact skin redness, swelling, the production of vesicles, scaling and thickening of the skin.

Alkyl alcohols of chain length C6-13 are absorbed from skin, when inhaled or swallowed but show evidence of little harm. They are broken down and rapidly excreted by the body.

The Branched Chain Saturated Alcohol (BCSA) group of fragrance ingredients was evaluated for safety. The fifteen materials tested have low acute toxicity. Following repeated application, seven materials had low whole-body toxicity.

In humans, no evidence of skin irritation was found at concentrations of 2-10%. Undiluted, 11 materials evaluated caused moderate to severe eye irritation. As current levels encountered during use are low, eye irritation is unlikely during routine use. The materials have no or low potential to cause sensitization. For individuals who are already sensitized, an elicitation reaction is possible. The BCSA are not expected to cause light-mediated toxicity or allergy.

Testing has not shown this group of materials to cause genetic toxicity. Whether this group has a cancer-promoting effect is unclear.

Acute Toxicity	✓	Carcinogenicity	×
Skin Irritation/Corrosion	✓	Reproductivity	✓
Serious Eye Damage/Irritation	✓	STOT - Single Exposure	✓
Respiratory or Skin sensitisation	×	STOT - Repeated Exposure	×
Mutagenicity	×	Aspiration Hazard	×

Legend:

Data either not available or does not fill the criteria for classification
 Data available to make classification

**SECTION 12 Ecological information** 

### Toxicity

	Endpoint	Test Duration (hr)	Species	Value
2-ethylhexanol	EC50	72h	Algae or other aquatic plants	11.5mg/l

Source

Chemwatch: **23004**Version No: **11.1.15.11** 

### Page 9 of 13

### 2-ETHYLHEXANOL

Issue Date: **15/04/2021**Print Date: **28/09/2021** 

EC50	48h	Crustacea	39mg/l	1
LC50	96h	Fish	>7.5mg/l	2
EC10(ECx)	72h	Algae or other aquatic plants	3.2mg/l	1

#### Legend:

Extracted from 1. IUCLID Toxicity Data 2. Europe ECHA Registered Substances - Ecotoxicological Information - Aquatic Toxicity 3. EPIWIN Suite V3.12 (QSAR) - Aquatic Toxicity Data (Estimated) 4. US EPA, Ecotox database - Aquatic Toxicity Data 5. ECETOC Aquatic Hazard Assessment Data 6. NITE (Japan) - Bioconcentration Data 7. METI (Japan) - Bioconcentration Data 8. Vendor Data

log Kow: 2.81 Ecotoxicology: Fish LC50 (48 h): Leuciscus idus 23 mg/l Biodegradability: >60% BOD of the ThOD (OECD 310F/ISO 9408) Toxicity invertebrate: ethyl formate some decomp in H2O.Rxn OH\*

Toxic to aquatic organisms.

Do NOT allow product to come in contact with surface waters or to intertidal areas below the mean high water mark. Do not contaminate water when cleaning equipment or disposing of equipment wash-waters.

Wastes resulting from use of the product must be disposed of on site or at approved waste sites.

May cause long-term adverse effects in the aquatic environment.

For alkyl alcohols C6-13

Environmental Fate: Biodegradation data for several alcohols in this group show the potential to biodegrade to a great extent within a 28-day period, and they are therefore not likely to persist in the environment. Data suggests that these alcohols are expected to partition primarily to water and soil, however their fate in air is interesting. The majority of the alcohols have relatively low Kow values which suggest that they will not readily be carried by suspended organic matter in air and precipitate to aquatic and terrestrial environments to a significant extent. The alkyl alcohols are not expected to hydrolyse at a measurable rate. Alkyl alcohols have the potential to volatilise, and may be transformed by atmospheric oxidation.

Ecotoxicity: The alkyl alcohols ranging from hexanol, branched and linear, to C11-14 iso-, C13 rich have been shown to produce an expected increasing level of acute toxicity to freshwater fish and invertebrates. A similar level of toxicity is expected for algae from the lower to higher molecular weight products. These alcohols demonstrate a moderate to high degree of aquatic toxicity from the low to high molecular weight products, respectively. Fish LC50 (96 h)97.7 mg/l34.5 mg/l14 mg/l10.1 mg/l3.1 mg/l1.2 mg/l0.42 mg/l. Data show the potential to cause acute toxicity in Daphnid. EL50/ EC50 (48 h) ranges from 137 to 0.7 with increasing molecular weight. Acute experimental toxicity threshold values are reported for freshwater alga (Scenendesmus quadricauda) and are used as read across data to the C8-10, iso, C9 rich alkyl alcohol product; this result suggests that C9-11, C10 rich, alkyl alcohol has the potential to cause acute toxicity (based on cell growth) at a concentration above 8.5 mg/l.

DO NOT discharge into sewer or waterways

#### Persistence and degradability

Ingredient	Persistence: Water/Soil	Persistence: Air
2-ethylhexanol	LOW	LOW

### Bioaccumulative potential

Ingredient	Bioaccumulation
2-ethylhexanol	LOW (BCF = 27)

# Mobility in soil

Ingredient	Mobility
2-ethylhexanol	LOW (KOC = 26.01)

# **SECTION 13 Disposal considerations**

# Waste treatment methods

Product / Packaging disposal

- ▶ Containers may still present a chemical hazard/ danger when empty.
- Return to supplier for reuse/ recycling if possible.

## Otherwise:

- If container can not be cleaned sufficiently well to ensure that residuals do not remain or if the container cannot be used to store the same product, then puncture containers, to prevent re-use, and bury at an authorised landfill.
- ▶ Where possible retain label warnings and SDS and observe all notices pertaining to the product.

Legislation addressing waste disposal requirements may differ by country, state and/ or territory. Each user must refer to laws operating in their area. In some areas, certain wastes must be tracked.

A Hierarchy of Controls seems to be common - the user should investigate:

- ► Reduction
- Reuse
- ► Recycling
- Disposal (if all else fails)

This material may be recycled if unused, or if it has not been contaminated so as to make it unsuitable for its intended use. If it has been contaminated, it may be possible to reclaim the product by filtration, distillation or some other means. Shelf life considerations should also be applied in making decisions of this type. Note that properties of a material may change in use, and recycling or reuse may not always be appropriate.

- DO NOT allow wash water from cleaning or process equipment to enter drains.
- It may be necessary to collect all wash water for treatment before disposal.
- In all cases disposal to sewer may be subject to local laws and regulations and these should be considered first.
- Where in doubt contact the responsible authority.
- Recycle wherever possible or consult manufacturer for recycling options.
- Consult State Land Waste Authority for disposal.
- ▶ Bury or incinerate residue at an approved site.
- Recycle containers if possible, or dispose of in an authorised landfill.

# **SECTION 14 Transport information**

# Labels Required

Issue Date: **15/04/2021**Print Date: **28/09/2021** 







# Land transport (DOT)

Land transport (DO1)			
UN number	3082		
UN proper shipping name	Environmentally hazardous substance, liquid, n.o.s. (contains 2-ethylhexanol)		
Transport hazard class(es)	Class 9 Subrisk Not Applicable		
Packing group	III		
Environmental hazard	Environmentally hazardous		
Special precautions for user	Hazard Label   9		

For Individual Packages of Environmentally Hazardous Substances meeting the descriptions of UN 3077 or UN 3082 that contain LESS THAN the reportable quantity (5000 lbs) - Not Regulated

For Individual Packages of Environmentally Hazardous Substances meeting the descriptions of UN 3077 or UN 3082 that contain MORE THAN the reportable quantity (5000 lbs) - Regulated and classified as below:

# Air transport (ICAO-IATA / DGR)

	<u>-</u>			
UN number	3082			
UN proper shipping name	Environmentally hazardo	ous substance, liquid, n.o.s. * (contains	2-ethylhexanol)	
Transport hazard class(es)	ICAO/IATA Class ICAO / IATA Subrisk ERG Code	9 Not Applicable 9L		
Packing group				
Environmental hazard	Environmentally hazardous			
Special precautions for user		Qty / Pack Packing Instructions	A97 A158 A197 A215 964 450 L 964 450 L Y964 30 kg G	

# Sea transport (IMDG-Code / GGVSee)

UN number	3082	3082				
UN proper shipping name	ENVIRONMENTALL	ENVIRONMENTALLY HAZARDOUS SUBSTANCE, LIQUID, N.O.S. (contains 2-ethylhexanol)				
Transport hazard class(es)		9 Not Applicable				
Packing group						
Environmental hazard	Marine Pollutant					
Special precautions for user	EMS Number Special provisions Limited Quantities					

# Transport in bulk according to Annex II of MARPOL and the IBC code

Product name	Pollution Category	Ship Type
Oxygenated aliphatic hydrocarbon mixture	Z	3

# Transport in bulk in accordance with MARPOL Annex V and the IMSBC Code

Product name

Issue Date: **15/04/2021**Print Date: **28/09/2021** 

Product name	Group
2-ethylhexanol	Not Available

# Transport in bulk in accordance with the ICG Code

Product name	Ship Type
2-ethylhexanol	Not Available

# **SECTION 15 Regulatory information**

# Safety, health and environmental regulations / legislation specific for the substance or mixture

# 2-ethylhexanol is found on the following regulatory lists

US - Massachusetts - Right To Know Listed Chemicals	
US DOE Temporary Emergency Exposure Limits (TEELs)	
US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory	

US TSCA Chemical Substance Inventory - Interim List of Active Substances

US TSCA Section 4/12 (b) - Sunset Dates/Status

# **Federal Regulations**

# Superfund Amendments and Reauthorization Act of 1986 (SARA)

### Section 311/312 hazard categories

Section 31/312 hazard categories	
Flammable (Gases, Aerosols, Liquids, or Solids)	Yes
Gas under pressure	No
Explosive	No
Self-heating	No
Pyrophoric (Liquid or Solid)	No
Pyrophoric Gas	No
Corrosive to metal	No
Oxidizer (Liquid, Solid or Gas)	No
Organic Peroxide	No
Self-reactive	No
In contact with water emits flammable gas	No
Combustible Dust	No
Carcinogenicity	No
Acute toxicity (any route of exposure)	Yes
Reproductive toxicity	Yes
Skin Corrosion or Irritation	Yes
Respiratory or Skin Sensitization	No
Serious eye damage or eye irritation	Yes
Specific target organ toxicity (single or repeated exposure)	No
Aspiration Hazard	No
Germ cell mutagenicity	No
Simple Asphyxiant	No
Hazards Not Otherwise Classified	No

# US. EPA CERCLA Hazardous Substances and Reportable Quantities (40 CFR 302.4)

None Reported

# State Regulations

# US. California Proposition 65

None Reported

# National Inventory Status

National inventory Status		
National Inventory	Status	
Australia - AIIC / Australia Non-Industrial Use	Yes	
Canada - DSL	Yes	
Canada - NDSL	No (2-ethylhexanol)	
China - IECSC	Yes	
Europe - EINEC / ELINCS / NLP	Yes	
Japan - ENCS	Yes	
Korea - KECI	Yes	
New Zealand - NZIoC	Yes	
Philippines - PICCS	Yes	
USA - TSCA	Yes	

 Chemwatch: 23004
 Page 12 of 13
 Issue Date: 15/04/2021

 Version No: 11.1.15.11
 2-ETHYLHEXANOL
 Print Date: 28/09/2021

National Inventory	Status
Taiwan - TCSI	Yes
Mexico - INSQ	Yes
Vietnam - NCI	Yes
Russia - FBEPH	Yes
Legend:	Yes = All CAS declared ingredients are on the inventory  No = One or more of the CAS listed ingredients are not on the inventory. These ingredients may be exempt or will require registration.

# **SECTION 16 Other information**

Revision Date	15/04/2021
Initial Date	28/06/2003

### **SDS Version Summary**

Version	Date of Update	Sections Updated
10.1.2.1	16/09/2016	Acute Health (inhaled), Acute Health (skin), Acute Health (swallowed), Appearance, Chronic Health, Toxicity and Irritation (Other), Use
11.1.2.1	14/04/2021	Expiration. Review and Update
11.1.3.1	10/05/2021	Regulation Change
11.1.4.1	24/05/2021	Regulation Change
11.1.4.2	30/05/2021	Template Change
11.1.4.3	04/06/2021	Template Change
11.1.4.4	05/06/2021	Template Change
11.1.4.5	09/06/2021	Template Change
11.1.4.6	11/06/2021	Template Change
11.1.4.7	14/06/2021	Template Change
11.1.4.8	05/07/2021	Template Change
11.1.5.8	14/07/2021	Regulation Change
11.1.6.8	15/07/2021	Regulation Change
11.1.6.9	01/08/2021	Template Change
11.1.7.9	02/08/2021	Regulation Change
11.1.8.9	05/08/2021	Regulation Change
11.1.9.9	09/08/2021	Regulation Change
11.1.10.9	16/08/2021	Regulation Change
11.1.10.10	29/08/2021	Template Change
11.1.11.10	30/08/2021	Regulation Change
11.1.12.10	06/09/2021	Regulation Change
11.1.13.10	13/09/2021	Regulation Change
11.1.13.11	16/09/2021	Template Change
11.1.14.11	16/09/2021	Regulation Change
11.1.15.11	23/09/2021	Regulation Change

## Other information

Classification of the preparation and its individual components has drawn on official and authoritative sources as well as independent review by the Chemwatch Classification committee using available literature references.

The SDS is a Hazard Communication tool and should be used to assist in the Risk Assessment. Many factors determine whether the reported Hazards are Risks in the workplace or other settings. Risks may be determined by reference to Exposures Scenarios. Scale of use, frequency of use and current or available engineering controls must be considered.

## **Definitions and abbreviations**

PC-TWA: Permissible Concentration-Time Weighted Average

PC-STEL: Permissible Concentration-Short Term Exposure Limit

IARC: International Agency for Research on Cancer

ACGIH: American Conference of Governmental Industrial Hygienists

STEL: Short Term Exposure Limit

TEEL: Temporary Emergency Exposure Limit。

IDLH: Immediately Dangerous to Life or Health Concentrations

ES: Exposure Standard
OSF: Odour Safety Factor

NOAEL :No Observed Adverse Effect Level

LOAEL: Lowest Observed Adverse Effect Level

TLV: Threshold Limit Value LOD: Limit Of Detection

OTV: Odour Threshold Value BCF: BioConcentration Factors BEI: Biological Exposure Index

AIIC: Australian Inventory of Industrial Chemicals

DSL: Domestic Substances List

Chemwatch: 23004 Page 13 of 13 Issue Date: 15/04/2021 Version No: 11.1.15.11 Print Date: 28/09/2021

2-ETHYLHEXANOL

NDSL: Non-Domestic Substances List

IECSC: Inventory of Existing Chemical Substance in China

EINECS: European INventory of Existing Commercial chemical Substances

ELINCS: European List of Notified Chemical Substances

NLP: No-Longer Polymers ENCS: Existing and New Chemical Substances Inventory

KECI: Korea Existing Chemicals Inventory

NZIoC: New Zealand Inventory of Chemicals PICCS: Philippine Inventory of Chemicals and Chemical Substances

TSCA: Toxic Substances Control Act TCSI: Taiwan Chemical Substance Inventory INSQ: Inventario Nacional de Sustancias Químicas

NCI: National Chemical Inventory

FBEPH: Russian Register of Potentially Hazardous Chemical and Biological Substances

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